

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
5 April 2001 (05.04.2001)

PCT

(10) International Publication Number  
**WO 01/23892 A1**

- (51) International Patent Classification<sup>7</sup>: G01N 33/543, C12Q 1/68, G01N 27/00, 29/02 Physical and Engineering Science, Guelph, Ontario N1G 2W1 (CA).
- (21) International Application Number: PCT/CA00/01139 (74) Agent: WOODLEY, John, H.; Sim & McBurney, 6th Floor, 330 University Avenue, Toronto, Ontario M5G 1R7 (CA).
- (22) International Filing Date:  
29 September 2000 (29.09.2000) (81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW.
- (25) Filing Language: English
- (26) Publication Language: English
- (30) Priority Data:  
60/156,714 30 September 1999 (30.09.1999) US (84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).
- (71) Applicant (*for all designated States except US*): SENSORCHEM INTERNATIONAL CORPORATION [CA/CA]; 170 College Street, Room 308A, Toronto, Ontario M5S 3E3 (CA).
- (72) Inventors; and
- (75) Inventors/Applicants (*for US only*): THOMPSON, Michael [CA/CA]; c/o University of Toronto, Department of Chemistry, 80 St. George Street, Toronto, Ontario M5S 3H6 (CA). HAYWARD, Gordon, L. [CA/CA]; c/o University of Guelph, School of Engineering, College of
- Published:  
— With international search report.
- For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.*

WO 01/23892 A1

(54) Title: TRAVERSE SHEAR MODE PIEZOELECTRIC CHEMICAL SENSOR

(57) Abstract: The present invention relates to a process for sensing biological or chemical changes in molecular structural shape or mass of molecules attached to the surface of a transverse shear piezoelectric oscillating molecular sensing device driven by a network analyzer. The process comprises the steps of i) exciting the sensor device at a series of predetermined frequencies, ii) collecting data to determine values for the predetermined parameters of series resonance frequency shift (fS), motional resistance (RM), motional inductance (LM), motional capacitance (CM), electrostatic capacitance (Co) and boundary layer slip parameter ( $\alpha$ ); and iii) determining relative changes in the measured parameters to detect thereby any changes in molecular structural shape or mass at sensing device surface.

## **TRAVERSE SHEAR MODE**

## **PIEZOELECTRIC CHEMICAL SENSOR**

### **Field of the Invention**

5           This invention relates to a process of detecting specific molecules in a liquid (the analyte) with receiving molecules, (the receptors) which are attached to the surface of a thickness shear mode acoustic sensor (TSM). Acoustic energy generated in the sensor is transferred to and from the fluid depending on the surface coupling behaviour. The coupling is altered when the analyte binds  
10   to the receptor producing easily measured changes in the electrical characteristics of the sensor.

          The invention further relates to the application of the measurement of the coupling effects to the sensing of biomolecules, and other molecules of biological significance such as drugs, in general. For example, the receptor may  
15   be a protein, a single oligonucleotide strand, DNA or RNA and the analyte a protein, drug or complementary strands of DNA or RNA. The interaction between the analyte and the sensor bound receptor can be identified through a quantitative TSM response. Other measurement scenarios are possible through the detection of changes in the acoustic coupling between the sensor surface  
20   and the surrounding liquid.

### **Background of the Invention**

          A TSM sensor is a device which generates mechanical vibrations from an electrical signal and uses these vibrations to detect and/or quantify particular chemical or biochemical substances present in a medium surrounding the  
25   sensor (the analyte). Acoustic energy is stored and dissipated both in the device itself, and through interfacial coupling, in a surrounding liquid medium. By coating the sensor with one or more layers of a substance which interacts with the analyte, the energy storage and transfer processes change when the interaction occurs. This changes the acoustic resonance of the sensor, which  
30   can be observed by measuring the electrical impedance of the sensor. The

applicants have published several papers in this field and they are listed as follows:

- 5           1)     F. Ferrante, A.L. Kipling and M. Thompson, "Molecular Slip At The Solid-Liquid Interface Of An Acoustic Wave Sensor", *J. Appl. Phys.* 76(6):3448-3462, 1994;
- 2)     G.L. Hayward and M. Thompson, "A Transverse Shear Model Of A Piezoelectric Chemical Sensor", *Amer. Inst. Physics* 83(40:2194-2201, 1998;
- 10           3)     Cavic B.A. et al., "Acoustic Waves And The Real-Time Study Of Biochemical Macromolecules At The Liquid/Solid Interface", *Faraday Discuss.* 107:159-176, 1997;
- 15           4)     H. Su and M. Thompson, "Rheological And Interfacial Properties Of Nucleic Acid Films Studies By Thickness-Shear Mode Sensor And Network Analysis", *Can. J. Chem.* 74:344-358, 1996.

20           There are several mechanisms whereby a TSM sensor responds to chemical change on its surface when it is immersed in a liquid. Surface mass deposition occurs when the analyte binds to the receptor on the sensor surface. This increases the storage of acoustic energy through the inertia of the added mass. Acoustic energy may also be stored through the elastic deformation of a  
25           coating on the surface. The elasticity of the coating may also change when the analyte binds to the receptor coating. These energy storage modes determine the resonant characteristics of the sensor which can easily be measured electrically. These processes are well known. Examples of piezoelectric sensors are described, for example in U.S. Patents 5,374,521 and 5,658,732.

Viscous loading occurs when acoustic energy is transferred to the liquid. Some of the acoustic energy is stored by the inertia of the fluid moving with the sensor surface and can be transferred back to the sensor, but acoustic energy is also dissipated by internal friction within the fluid. The viscous loading effect is also well known, however in the current use of this effect, the transfer of acoustic energy at the surface is considered to be perfect, that is, there is no slip between the sensor surface and the adjacent fluid molecules.

The current practice is based on the well known Butterworth - van Dyke model of a piezoelectric resonator which consists of a resistor, inductor and capacitor in series, all in parallel with another capacitor. The series arm of this network is called the motional arm. Further details of this model and the calculation of the following parameters may be found in the above paper entitled "Rheological and Interfacial Properties of Nucleic Acid Films Studies by Thickness-Shear Mode Sensor and Network Analysis".

15

#### Motional Inductance

The motional inductance,  $L_M$ , represents the inertial energy stored by the sensor. It depends on the mass of the TSM sensor as well as the mass of material (the analyte) added to the surface. Since liquid coupled to the surface can store and return acoustic energy,  $L_M$  is also dependent on the viscosity of the liquid.

20

#### Motional Resistance

The motional resistance,  $R_M$ , is intrinsically related to the energy dissipated by the sensor.

25

Accordingly, any imposition of material (or loss of material) that has a viscous property or changes in the viscosity of the liquid will result in a change in the energy dissipation and hence  $R_M$ .

### Motional Capacitance

The motional capacitance,  $C_M$ , represents the elastic energy stored by the sensor. The absorption or chemical binding of the analyte to the coating can have a large effect on the viscoelastic properties of the coating. Depending on the thickness, an added (or removed) layer of material may change the elasticity of the sensor and thus affect  $C_M$ . Although most fluids are considered to be viscous, at the high frequencies used in piezoelectric quartz sensors, the liquid may also have elastic properties.

### 10 Static Capacitance

The static capacitance  $C_0$  represents the dielectric constant of the quartz, but includes that of the medium through the electric field. Charge interactions between the analyte and the sensor coating will affect this value.

### Summary of the Invention

15 According to an aspect of the invention, there is provided a process for sensing biological or chemical changes in molecular structural shape or mass of molecules attached to the surface of a transverse shear piezoelectric oscillating molecular sensing device driven by a network analyzer, said process comprising:

- 20 i) exciting said sensor device at a series of predetermined frequencies;
- ii) collecting data to determine values for the predetermined parameters of series resonance frequency shift ( $f_S$ ), motional resistance (RM), motional inductance (LM), motional capacitance (CM), electrostatic capacitance ( $C_0$ ) and boundary layer slip parameter ( $\alpha$ ); and
- 25 iii) determining relative changes in said measured parameters to detect thereby any changes in molecular structural shape or mass at sensing device surface.

In accordance with another aspect of the invention there is provided a method of determining the efficiency of acoustic coupling between a sensor and the surrounding fluid, said method comprising:

30

- a) applying an electrical signal of known frequency and voltage to the sensor;
- b) measuring the current through the sensor to determine the impedance at the known frequency;
- 5 c) repeating steps a) and b) over a range of frequencies to generate a set of impedance data; and
- d) fitting the measured impedance data to determine an  $\alpha$  parameter which represents coupling strength.

#### **Detailed Description of the Preferred Embodiments**

10 This invention is based on the measurement of phenomena based on imperfect acoustic coupling between the sensor surface and the liquid. The nature of this coupling determines the strength of the viscous loading and elastic effects depending on such parameters as the surface free energy and the molecular conformation of the sensor coating. These molecular parameters are  
15 very sensitive to chemical changes at the surface and therefore acoustic coupling provides a novel sensing mechanism.

The impedance measurements are carried out by applying an electrical signal of known frequency and voltage to the sensor and measuring the current through the sensor. Through Ohm's law, this provides the impedance at the  
20 known frequency. By performing this measurement over a range of frequencies, a set of data is generated. The above described, specifically selected parameters of  $L_M$ ,  $R_M$ ,  $C_M$  and  $C_O$  have been found to be the determining parameters for indicating a mass or conformation change at the TSM surface. Hence these parameters are fitted to the data.

25 While the Butterworth - van Dyke model provides useful information, it is an electrical analogy which presents the information unclearly. An alternate model of the TSM sensor is based on a solution of the equations of motion and electric fields. With this second model as set out in the aforementioned paper entitled "Molecular Slip At The Solid-Liquid Interface Of An Acoustic Wave  
30 Sensor" and "A Transverse Shear Model Of A Piezoelectric Chemical Sensor", the deposited mass and the coupling may be determined directly by fitting the

electrical impedance data obtained as above. The coupling is represented by a slip parameter,  $\alpha$ , which arises from a slip boundary condition used in solving the set of equations. The common approach is to assume perfect coupling and to set  $\alpha = 1$ . In this invention,  $\alpha$  is taken to be a complex number which is  
5 determined by fitting the measured impedance data.

The sensing process is understood to be occurring at the interface between the solid device and the liquid medium. Ligands for biological macromolecules include small molecules, ions, proteins, peptides, and strands of both DNA and RNA. The interaction of these entities with the biological  
10 molecules attached to the sensor will cause an alteration of the physical properties of the film resulting, in turn, in changes in the measured parameters. These changes will very clearly result from a combination of some or all of the above response mechanisms particular for each chemical situation. In this regard, the dimensions of the newly bound ligand is an important consideration.

15 The signaling species coated onto the acoustic biosensor are proteins (antibodies, enzymes, hormones, molecular receptors, etc.) and nucleic acids (oligonucleotides, DNA and RNA) attached to the device surface. These molecules exist in a highly hydrated form which can be considered to constitute very viscous gels.

20 The effect of viscous loading is the result of acoustic energy transfer to and from the surrounding medium. This in turn depends on the nature of the contact between the surface and the medium. The contact is controlled by such chemical properties as hydrogen bonding, dispersion interactions and interfacial charge. The process can be viewed as a drag existing between the surface  
25 coating and the liquid.  $\alpha$  represents the coupling strength but also contains phase shift information. This provides additional information regarding relative mass of liquid molecules compared to those of the sensor surface and when correlated with the selected Butterworth – van Dyke model provide a determination on what is happening at the TSM surface, namely, mass and/or  
30 molecular structural shift or change in conformation.

### Example

The human immunodeficiency virus type I (HIV-I) is strongly regulated at the transcriptional level by the interaction of an 86-amino acid protein, Tat, with the trans activation responsive element at the 5' -end of the viral messenger RNA transcript (TAR). The TAR-Tat system is an important target for drug discovery research because the binding of the regulatory protein to TAR can be blocked by small molecules.

In this application we compute the slip parameter  $\alpha$ , for the binding of Tat-derived peptides to TAR immobilized on a sensor surface. The TAR RNA is synthesized, with a biotin moiety at the 5' -end, on a DNA synthesizer by standard phosphoramidite chemistry. The acoustic wave sensor is incorporated into a flow-through configuration and electrically connected to an acoustic network analyzer. A dispersion of 100-500  $\mu$ l of the reagent neutravidin is injected into the apparatus and the protein adsorbs to the gold electrode surface of the acoustic wave sensor. A dispersion of biotinylated TAR- RNA (100-500  $\mu$ l) is introduced into the system where the formation of the biotin-avidin complex results in attachment of TAR to the sensor surface. Various Tat-derived peptides are then introduced into the flow-trough system. In this particular application the following peptides are specified: tat<sub>12</sub>, tat<sub>20</sub>, and tat<sub>30</sub> where the subscript refers to the number of amino acids in the peptide. Dispersions of peptide (100-500  $\mu$ l) are injected into the system. On binding of peptide to TAR in real time transient responses in the aforementioned parameters are obtained. The computed  $\alpha$  parameter for the various responses, which distinguishes the nature on binding, are as follows:

Tat<sub>12</sub> baseline 1.978 @20.85 degrees  
signal 1.964 @ 20.97 degrees

Tat<sub>20</sub> baseline 1.985 @21.42 degrees  
signal 1.926@ 18.15 degrees



Tat<sub>30</sub> baseline 1.982 @ 22.61 degrees  
signal 1.994 @ 23.03 degrees

5 Tat<sub>12</sub> displays a small decrease in slip magnitude with an increase in phase,  
whereas tat<sub>20</sub> shows large decreases in magnitude and phase. Tat<sub>30</sub> depicts  
smaller increase in magnitude and phase.

Although preferred embodiments of the invention have been described  
herein in detail, it will be understood by those skilled in the art that variations  
may be made thereto without departing from the spirit of the invention or the  
10 scope of the appended claims.

**CLAIMS:**

1. A process for sensing biological or chemical changes in molecular structural shape or mass of molecules attached to the surface of a transverse  
5 shear piezoelectric oscillating molecular sensing device driven by a network analyzer, said process comprising:
  - i) exciting said sensor device at a series of predetermined frequencies ;
  - 10 ii) collecting data to determine values for the predetermined parameters of series resonance frequency shift (fS), motional resistance (RM), motional inductance (LM), motional capacitance (CM), electrostatic capacitance (Co) and boundary layer slip parameter ( $\alpha$ ); and
  - 15 iii) determining relative changes in said measured parameters to detect thereby any changes in molecular structural shape or mass at sensing device surface.
2. The process according to claim 1 further comprising the step of:
  - 20 iv) correlating said changes with a calibrated set of data for said parameters to determine a value for change in molecular conformation and/or molecular mass.
3. The process according to claim 1 wherein a change in slip parameter ( $\alpha$ ) and an essentially zero change in series resonant frequency shift confirms a change in molecular structural shape and essentially zero change in mass.  
25
4. The process according to claim 1 wherein said changes in molecular mass or conformation are generated by an interaction between entities bound to the sensor and molecules in the surrounding liquid medium.
- 30 5. The process according to claim 4 wherein said entities bound to the sensor are selected from the group consisting of proteins and nucleic acids.

6. The process according to claim 5 wherein said proteins are selected from the group consisting of antibodies, enzymes, molecular receptors, receptor ligands and polypeptides.
- 5 7. The process according to claim 5 wherein said nucleic acids are selected from the group consisting of DNA, RNA and oligonucleotides.
8. The process according to claim 4 wherein said molecules in liquid medium are selected from the group consisting of proteins and nucleic acids.
- 10 9. The process according to claim 8 wherein said proteins are selected from the group consisting of antibodies, enzymes, molecular receptors, receptor ligands and polypeptides.
- 15 10. The process according to claim 8 wherein said nucleic acids are selected from the group consisting of DNA, RNA and oligonucleotides.
11. A method of determining the efficiency of acoustic coupling between a sensor and the surrounding fluid, said method comprising:
- 20 a) applying an electrical signal of known frequency and voltage to the sensor;
- b) measuring the current through the sensor to determine the impedance at the known frequency;
- c) repeating steps a) and b) over a range of frequencies to
- 25 generate a set of impedance data; and
- d) fitting the measured impedance data to determine an  $\alpha$  parameter which represents coupling strength.
12. The method according to claim 11, wherein the  $\alpha$  parameter is other
- 30 than 1.

13. The method according to claim 11 wherein the magnitude of said  $\alpha$  parameter is dependant on molecular mass and/or molecular conformation at the sensor surface.

## INTERNATIONAL SEARCH REPORT

International Application No

PCT/CA 00/01139

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 G01N33/543 C12Q1/68 G01N27/00 G01N29/02

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 G01N C12Q

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, INSPEC, WPI Data, PAJ

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US 5 374 521 A (KIPLING ARLIN L ET AL) 20 December 1994 (1994-12-20) cited in the application column 1, line 66 - column 2, line 3 column 2, line 40 - line 46 column 3, line 23 - line 45 column 5, line 10 - line 21 column 5, line 43 - line 46 column 5, line 59 - line 66 --- --/--	1-13

☒ Further documents are listed in the continuation of box C.☒ Patent family members are listed in annex.

## \* Special categories of cited documents:

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*E\* earlier document but published on or after the international filing date
- \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the international filing date but later than the priority date claimed

- \*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- \*X\* document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- \*Y\* document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- \*&\* document member of the same patent family

Date of the actual completion of the international search

11 January 2001

Date of mailing of the international search report

19/01/2001

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2  
 NL - 2280 HV Rijswijk  
 Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,  
 Fax: (+31-70) 340-3016

Authorized officer

Joyce, D

## INTERNATIONAL SEARCH REPORT

International Application No

PCT/CA 00/01139

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	FERRANTE F ET AL: "MOLECULAR SLIP AT THE SOLID-LIQUID INTERFACE OF AN ACOUSTIC-WAVE SENSOR" JOURNAL OF APPLIED PHYSICS, US, AMERICAN INSTITUTE OF PHYSICS. NEW YORK, vol. 76, no. 6, 15 September 1994 (1994-09-15), pages 3448-3462, XP000470063 ISSN: 0021-8979 cited in the application column 6, line 12 - line 14 column 16, paragraph 1 ----	1-13
A	US 5 306 644 A (MYERHOLTZ CARL A ET AL) 26 April 1994 (1994-04-26) column 5, line 22 - line 43 ----	4-10
A	US 4 735 906 A (BASTIAANS GLENN J ET AL) 5 April 1988 (1988-04-05) abstract column 8, line 63 - line 65 ----	1,2,5,6
A	US 5 658 732 A (EBERSOLE RICHARD CALVIN ET AL) 19 August 1997 (1997-08-19) cited in the application column 5, line 2 - line 18 column 6, line 24 - line 37 ----	1,5,6
A	PATENT ABSTRACTS OF JAPAN vol. 1998, no. 10, 31 August 1998 (1998-08-31) & JP 10 115648 A (ADVANTEST CORP), 6 May 1998 (1998-05-06) abstract -----	1

# INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/CA 00/01139

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 5374521 A	20-12-1994	CA 2078184 A,C	18-03-1993
US 5306644 A	26-04-1994	US 5130257 A	14-07-1992
		US 5283037 A	01-02-1994
		DE 4409588 A	06-10-1994
		JP 6308127 A	04-11-1994
		EP 0361729 A	04-04-1990
		JP 2122242 A	09-05-1990
		DE 69027077 D	27-06-1996
		DE 69027077 T	02-10-1996
		EP 0416730 A	13-03-1991
		JP 3100438 A	25-04-1991
		EP 0542469 A	19-05-1993
		JP 5240762 A	17-09-1993
US 4735906 A	05-04-1988	NONE	
US 5658732 A	19-08-1997	CA 2066643 A	05-04-1991
		DE 69010506 D	11-08-1994
		DE 69010506 T	22-12-1994
		EP 0494896 A	22-07-1992
		JP 5500715 T	12-02-1993
		WO 9105261 A	18-04-1991
JP 10115648 A	06-05-1998	NONE	